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Convenient synthesis of *cis-O*-isopropylidene-3,5-cyclohexadien-1,2-diol

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Abstract—The title compound was obtained by a convenient and scaleable three-step procedure, starting from the readily available and relatively inexpensive *myo*-inositol. The key-step of the route is an unprecedented tandem reductive elimination of two *vic*-dimesylate moieties to a conjugated diene.

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cis-3,5-Cyclohexadien-1,2-diol **2** and its protected analogues are valuable starting materials for the synthesis of a great number of biologically active substances,^{1,2} unusual molecules³ and high-performance materials.⁴ The great interest in these substances arises from the presence of two easily derivatized functional groups (olefines and hydroxyls) in a specific arrangement, that allows a great number of chemical transformations.^{1,5} Disappointedly, the high cost and the variable quality of commercial **2** is a serious drawback for large-scale studies and applications. Biological oxidation of benzene and its derivatives, by the use of various mutant strains of *Pseudomonas putida*^{4f,6} (Scheme 1), is advantageous, but this does require large and specialist equipment, often unfamiliar to organic chemists, and only affords limited amounts of material.

A convenient four-step, 26% overall yield, procedure for the synthesis of isopropylidene or benzylidene protected *cis*-1,2-dihydrocatechol on a large scale was proposed by Yang and co-workers.^{30,p} This procedure noticeably reduces the cost for the preparation of protected *cis*-1,2dihydrocatechols, despite the use of the relatively expensive 1,4-cyclohexadiene as starting material. A four-step, 24% overall yield, synthesis of cyclohexylidene protected



Scheme 1.

3,5-cyclohexadien-1,2-diol was also reported by Mereyala and Pannala,⁷ starting from *myo*-inositol.

In this letter, we describe a convenient and scaleable three-step synthesis of *cis*-3,5-cyclohexadien-1,2-diol acetonide, that is one of the more widely used protected dihydrocatechols, starting from *myo*-inositol and other readily available and inexpensive reagents. The starting material *myo*-inositol **3** was protected with 2,2-dimeth-oxypropane, according to the procedure of Gigg, to afford the acetal **4** in 70% yield (Scheme 2).⁸





Keywords: 1,2-Dihydrocatechol; Inositol; Conjugated dienes; Cycloadditions; Reductions.

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A number of methodologies were studied in order to transform *vic*-diols into olefins.⁹ In early attempts we looked for direct protocols to transform the four hydroxyls into the two conjugate double bonds. Disappointedly, the use of triphenylphosphine and iodine^{7,9,10} invariably led to the formation of tars. Tetrol **4** was hence transformed into tetramesyl-derivative **5**, in nearly quantitative yield, with a slight excess of methanesulfonyl chloride in pyridine (Scheme 3).¹¹



Scheme 3.

Many attempts to react mesylate **5** with sodium naphthalenide in dry THF at low temperature, 9,12 led invariably to a complex mixture of products. Mesylate **5** was refluxed 24 h in DMF with potassium iodide and zinccopper couple, 9,13 to afford the diene **6** in 20–46% concomitantly with variable amounts of cycloadduct **7** (Scheme 4).¹⁴





The quality of the zinc–copper couple¹⁵ was found to be crucial for the success of the reaction, in fact freshly activated metal afforded the higher yields of **6**. A further improvement in the synthesis was the use of NMP (*N*-methylpyrrolidinone) as solvent; this allowed the co-distillation at reduced pressure. This procedure allowed improved yields (48–60%); possibly because of the more efficient activation of the de-mesylation at the higher temperatures (145 °C), and the removal of the product from the reaction mixture. The absence of zinc salts in the crude mixture and the higher purity of **6** after extractions are other significant advantages of this procedure (dimer **7** was not observed in the product).

To conclude, in this letter we describe a new, more convenient, effective, three-step synthesis of 1,2-dihydrocatechol acetonide **6**, with an overall 36% yield. The key-step of the procedure is the unprecedented reductive elimination of a *vic*-tetramesylate to a conjugate diene.

Experimental

Synthesis of 1,2-*O*-isopropylidene-3,4,5,6-tetra-(methanesulfonyl)-*myo*-inositol (5)

Mesyl chloride (57.0 mL, 84.9 g, 741 mmol) was added via syringe to a solution of 4 (32.6 g, 148 mmol) and DMAP (1.0 g, 8.2 mmol) in pyridine (250 mL) maintained at 0 °C under argon. The mixture was left to rise to rt overnight and was poured onto crushed ice (800 g). The resulting white-off solid was filtered, washed with cold water (3×100 mL), *i*-PrOH (2×100 mL) and dried in vacuum over P_2O_5 to obtain 76.4 g (97% yield) of 5, pure enough for the next step. An analytical sample was obtained by recrystallisation from DMF/Et₂O, mp 210 °C. IR (KBr) v_{max} 3037, 2983, 2957, 1362, 1181, 973, 875, 818, 518 cm⁻¹; ¹H NMR (300 MHz, DMSO d_6) δ 5.37–5.28 (1H, m), 5.21–5.08 (2H, m), 4.87–4.76 (1H, m), 4.68–4.60 (1H, m), 4.51–4.43 (1H, m), 3.35 (3H, m), 3.32 (3H, m), 3.29 (3H, m), 3.28 (3H, m), 1.56 (3H, m), 1.35 (3H, m); ¹³C NMR (75 MHz, DMSO- d_6) δ 110.9, 80.7, 75.7, 74.8, 74.5, 73.8, 73.6, 39.2, 39.0, 38.7, 38.1, 27.0, 25.6. Anal. Calcd for C13H24O14S4: C, 29.32; H, 4.54. Found: C, 29.35; H, 4.51.

Synthesis of *cis-O*-isopropylidene-3,5-cyclohexadien-1,2diol (6)

In a 500 mL two necked round-bottomed flask equipped with a mechanical stirrer and distillation apparatus, a mixture of 5 (10.0 g, 18.78 mmol) and KI (24.0 g, 144.6 mmol) in NMP (200 mL) was heated at 120 °C with an oil bath. Approx. 30 mL of volatile materials were distilled off in vacuum (10 Torr) in 1 h.¹⁶ Freshly prepared Zn/Cu couple¹⁵ (from 15 g of Zn dust, 225 mmol) was added to the solution and the resulting mixture was heated at 120 °C for 24 h. Vacuum (20 Torr) was applied to the apparatus, the oil bath was heated at 145 °C and the volatile materials were distilled (bp 90-100 °C). The distillate was poured into a 3:2 satd aq NaCl/H₂O mixture (500 mL) and extracted with AcOEt $(3 \times 50 \text{ mL})$. The combined organic extracts were washed with water $(5 \times 50 \text{ mL})$, saturated NaCl $(3 \times 50 \text{ mL})$, dried over MgSO₄ and carefully concentrated at reduced pressure to afford 1.37-1.70 g (48-60% yield) of **6**, identical to authentic sample.

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References and notes

- (a) Hudlicky, T.; Entwistle, D. A.; Pitzer, K. K.; Thorpe, A. J. Chem. Rev. 1996, 96, 1195–1220; (b) Hudlicky, T. Chem. Rev. 1996, 96, 3–30.
- (a) Baran, A.; Kazaz, C.; Secen, H. *Tetrahedron* 2004, 60, 861–866; (b) Elango, S.; Yan, T.-H. J. Org. Chem. 2002, 67, 6954–6959; (c) Spiegel, D. A.; Njardarson, J. T.; Wood, J. L. *Tetrahedron* 2002, 58, 6545–6554; (d)

Sanfilippo, C.; Nicolosi, G. Tetrahedron: Asymmetry 2002, 13, 1889–1892; (e) Noguchi, H.; Aoyama, T.; Shioiri, T. Heterocycles 2002, 58, 471–474; (f) Yoshizaki, H.; Bäckvall, J.-E. J. Org. Chem. 1998, 63, 9339–9341; (g) Sanfilippo, C.; Patti, A.; Piattelli, M.; Nicolosi, G. Tetrahedron: Asymmetry 1997, 8, 1569–1573; (h) Patti, A.; Sanfilippo, C.; Piattelli, M.; Nicolosi, G. J. Org. Chem. 1996, 61, 6458–6461; (i) Patti, A.; Sanfilippo, C.; Piattelli, M.; Nicolosi, G. 7, 2665–2670; (j) Maras, A.; Secen, H.; Sutbeyaz, Y.; Balci, M. Turk. J. Chem. 1996, 20, 341–344.

- 3. (a) Noh, T.; Yu, H.; Jeong, Y.; Jeon, K.; Kang, S. J. Chem. Soc., Perkin Trans. 1 2001, 1066-1071; (b) Gasyna, Z.; Chen, G.; Gleiter, R.; Yang, N.-C. J. Am. Chem. Soc. 2000, 122, 12098-12111; (c) Reddy, G. D.; Wiest, O.; Hudlicky, T.; Schapiro, V.; Gonzalez, D. J. Org. Chem. 1999, 64, 2860-2863; (d) Noh, T.; Jeon, K.; Jeong, Y.; Jang, S.; Min, K. S. J. Chem. Soc., Perkin Trans. 2 1999, 1299–1304; (e) Grimme, W.; Wortmann, J.; Frowein, D.; Lex, J.; Chen, G.; Gleiter, R. J. Chem. Soc., Perkin Trans. 2 1998, 1893-1900; (f) Noh, T.; Gan, H.; Halfon, S.; Hrnjez, B. J.; Yang, N.-C. C. J. Am. Chem. Soc. 1997, 119, 7470-7482; (g) Wagaman, M. W.; Bellmann, E.; Cucullu, M.; Grubbs, R. H. J. Org. Chem. 1997, 62, 9076-9082; (h) Cossu, S.; Fabris, F.; De Lucchi, O. Synlett 1997, 1327-1334; (i) Herb, T.; Gleiter, R. Angew. Chem., Int. Ed. 1996, 35, 2368; (j) Pu, L.; Grubbs, R. H. J. Org. Chem. 1994, 59, 1351-1353; (k) Gan, H.; King, J. L.; Yang, N.-C. C. Tetrahedron Lett. 1989, 30, 1205-1208; (1) Yang, N. C.; Noh, T.; Gan, H.; Halfon, S.; Hrnjez, B. J. J. Am. Chem. Soc. 1988, 110, 5919-5920; (m) Yang, N.-C. C.; Yang, X.-Q. J. Am. Chem. Soc. 1987, 109, 3804-3805; (n) Yang, N.-C. C.; Horner, M. G. Tetrahedron Lett. 1986, 27, 543-546; (o) Yang, N. C. C.; Chen, M. J.; Chen, P. J. Am. Chem. Soc. 1984, 106, 7310-7315; (p) Yang, N. C. C.; Chen, M. J.; Chen, P.; Mak, K. T. J. Am. Chem. Soc. 1982, 104, 853-855.
- (a) Uno, H.; Ishikawa, T.; Hoshi, T.; Ono, N. Tetrahedron Lett. 2003, 44, 5163–5166; (b) Wagaman, M. W.; Grubbs, R. H. Macromolecules 1997, 30, 3978–3985; (c) Gin, D. L.; Conticello, V. P.; Grubbs, R. H. J. Am. Chem. Soc. 1994, 116, 10507–10519; (d) Kim, H. K.; Ober, C. J. Macromol. Sci., Pure Appl. Chem. 1993, A30, 877–897; (e) Conticello, V. P.; Gin, D. L.; Grubbs, R. H. J. Am. Chem. Soc. 1992, 114, 9708–9710; (f) Ballard, D. G. H.; Courtis, A.; Shirley, I. M.; Taylor, S. C. Macromolecules 1988, 21, 294–304.
- (a) Banwell, M. G.; Chun, C.; Edwards, A. J.; Voegtle, M. M. Aust. J. Chem. 2003, 56, 861–869; (b) Donohoe, T. J.; Blades, K.; Moore, P. R.; Waring, M. J.; Winter, J. J. G.; Helliwell, M. J. Org. Chem. 2002, 67, 7946–7956; (c) Newcombe, N. J.; Stemp, G. J. Org. Chem. 2002, 67, 7946–7956; (d) Hudlicky, T.; Gonzalez, D.; Gibson, D. T. Aldrichim. Acta 1999, 32, 35; (e) Han, X.; Khedekar, R. N.; Masnovi, J.; Baker, R. J. J. Org. Chem. 1999, 64,

5245–5250; (f) Banwell, M. G.; Haddad, N.; Hudlicky, T.; Nugent, T. C.; Mackay, M. F.; Richards, S. L. J. Chem. Soc., Perkin Trans. 1 1997, 1779–1791; (g) Donohoe, T. J.; Moore, P. R.; Beddoes, R. L. J. Chem. Soc., Perkin Trans. 1 1997, 43–51; (h) Roberts, S. M.; Sutton, P. W.; Wright, L. J. Chem. Soc., Perkin Trans. 1 1996, 1157–1166.

- (a) Johnson, R. A. Org. React. 2004, 63, 117–264; (b) Hudlicky, T.; Stabile, M. R.; Gibson, D. T.; Whited, G. M. Org. Synth., Coll. Vol. 2004, 10, 2340–2346; (c) Boyd, D. R.; Sharma, N. D.; Barr, S. A.; Dalton, H.; Chima, J.; Whited, G.; Seemayer, R. J. Am. Chem. Soc. 1994, 116, 1147–1148; (d) Patel, T. R.; Gibson, D. T. J. Bacteriol. 1976, 128, 842–850; (e) Gibson, D. T.; Hensley, M.; Yoshioka, H.; Mabry, T. J. Biochemistry 1970, 9, 1626– 1629; (f) Gibson, D. T.; Koch, J. R.; Kallio, R. E. Biochemistry 1968, 7, 2653–2658.
- Mereyala, H. B.; Pannala, M. Tetrahedron Lett. 1995, 36, 2121–2124.
- 8. Gigg, J.; Warren, C. D. J. Chem. Soc. (C) 1969, 2367–2371, In order to enhance the yields of 4 and to avoid the use of large amounts of ethanol for the extraction and recrystallisation, some slight modifications of the procedure was introduced: the mother liquors of the crude mixture of unreacted myo-inositol and 4, containing a mixture of diastereomeric diacetonides, were concentrated, dissolved in DCM, treated with a catalytic amount of 1 M HCl and stirred at rt for 18 h to afford a further crop of crude 4. Collected solid materials were extracted in a Soxelet apparatus for 48 h with ethanol (7.5 mL/g of starting myo-inositol). The cooled solution afforded 4 as colourless crystals.
- 9. Block, E. Org. React. 1984, 30, 457-566.
- Oscarson, S.; Svahnberg, P. Carbohydr. Res. 1998, 309, 207–212; Lange, G. L.; Gottardo, C. J. Org. Chem. 1995, 60, 2183–2187; Nagel, A. A.; Vincent, L. A. J. Org. Chem. 1982, 47, 4796–4799; Jarosz, S.; Hicks, D. R.; Fraser-Reid, B. J. Org. Chem. 1982, 47, 935–940.
- 11. Martin, R. L.; Norcross, B. E. J. Org. Chem. 1975, 40, 523–524.
- 12. Robins, M. J.; Lewandowska, E.; Wnuk, S. F. J. Org. Chem. 1998, 63, 7375–7381.
- Morimoto, Y.; Shirahama, H. *Tetrahedron* 1997, 53, 2013–2024; Yasui, K.; Tamura, Y.; Nakatani, T.; Kawada, K.; Ohtani, M. J. Org. Chem. 1995, 60, 7567–7574.
- 14. Rapid dimerization of 6 to 7 was observed to occur at rt when aliphatic solvents (e.g., pentane, hexanes) were used for the extractions. The use of ethyl acetate or toluene can prevent this phenomenon. Diene 6 resulted also stable for prolonged times at rt in DMF or NMP solutions.
- Smith, R. D.; Simmons, H. E.; Parham, W. E.; Bhavsar, M. D. Org. Synth. Coll. Vol. 1973, 5, 855–858.
- 16. Distillation allowed the removal of water, which was eventually present in the reagents and solvent, that were not previously dried.